



Excitonic and electronic transitions in Me–Sb₂Se₃ structures

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Abstract

The optical anisotropy of the Sb₂Se₃ crystals was investigated at 300 and 11 K. Excitonic features of four excitons (A, B, C, and D) were observed in the optical spectra of the Sb₂Se₃ single crystals and in the photoelectric spectra of the Me–Sb₂Se₃ structures. The exciton parameters, such as the ground ($n = 1$) and excited ($n = 2$) state positions and the binding energy (R_y), were determined. The effective mass of the electrons at the bottom of the conduction band ($m_c^* = 0.67m_0$) as well as the holes at the four top valence bands ($m_{v1}^* = 3.32m_0$, $m_{v2}^* = 3.83m_0$, $m_{v3}^* = 3.23m_0$ and $m_{v4}^* = 3.23m_0$) were calculated in the Γ -point of the Brillouin zone. The magnitude of the valence band splitting V_1 – V_2 due to the spin–orbit interaction ($\Delta_{so} = 35$ meV) and the crystal field ($\Delta_{cf} = 13$ meV) were estimated in the Brillouin zone center. The energy splitting between the bands V_3 – V_4 was 191 meV. The identified features were discussed based on both the theoretically calculated energy band structure and the excitonic band symmetry in the Brillouin zone ($k = 0$) for crystals with an orthorhombic symmetry ($Pnma$). The photoelectric properties of the Me–Sb₂Se₃ structures were investigated in the spectral range 1–1.8 eV under $E||c$ and $E\perp c$ polarization conditions and at different applied voltages.

Introduction

Antimony selenide (Sb₂Se₃) is an inorganic semiconductor compound with interesting photoelectric properties. This material has a high absorption coefficient ($\approx 10^5$ cm⁻¹) in the region of maximum solar energy radiation [1,2] which is corroborated by a 6.5% rapid increase in solar cell efficiency when Sb₂Se₃ is present [3–5]. Interestingly, this high absorption coefficient is 10³ times higher than the absorption in silicon [5–7] and encompasses a wide portion of the spectrum ranging from 1.0 eV to

2–3 eV. The crystalline structure of Sb₂Se₃ is quite uniform and stable which minimizes the energy loss due to radiation [3,7,8]. In combination, the binary arrangement (Sb, Se), high crystalline stability, low toxicity and low deposition temperature (melting point ≈ 611 °C) reduce the production costs [3–10]. It has been shown that Sb₂Se₃ has many applications in photovoltaic devices and thermoelectric systems where it can be used as a thin film [11], in thermovoltaic and switch devices [12], in