

Evidence of SnSb_2Te_4 rocksalt metastable phase of thin films grown by pulsed laser deposition

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SnSb_2Te_4 , an intermetallic compounds in the pseudobinary $\text{SnTe-Sb}_2\text{Te}_3$, looks as a promising candidate for phase-change material in non-volatile memories applications. Thin films having thickness from 50 to 150 nm were grown by pulsed laser deposition using SnSb_2Te_4 as structure of 21 *R*-type structure (*R-3m*) consisting of rocksalt-type blocks separated by van der Waals gaps [1], the thin films obtained had a simple NaCl-structure (*Fm-3m*) (metastable phase) as it was deduced from conventional XRD measurements in grazing incidence of the investigated films. The metastable phase transitioned to the stable complex structure during the measurements of electrical resistance on temperature dependence. $^{119\text{m}}\text{Sn}$ conversion electron Mössbauer spectroscopy was used to investigate the local environment of Sn atoms in as made film. The existence of Sn(II) in a high symmetry site was found as expected for SnSb_2Te_4 . Nevertheless, the dominant contribution was Sn(IV) attributed to SnO_2 phase, evidencing an oxidation of the films surface.

The evidence of the metastable NaCl-structure of GeSb_2Te_4 films induced by heating the amorphous film by laser irradiation were reported [2-4]. Nevertheless, reports about SnSb_2Te_4 film to date were not found.

Keywords: crystal structure, thin films, metastable phase

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Thin films of porphyrin-chalcogenide nanocomposite for methane sensing at room temperature

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Methane is one of the components of natural gas and has multiple applications. It is an odorless gas, flammable and combustible in the presence of oxygen. Its lower explosion limit (LEL) is 5.0 vol.%, which corresponds to 50 000 ppm, while the upper explosion limit (UEL) is 15 vol% [1]. The classical and versatile SnO₂ gas sensing material was initially developed by researchers from Japan, several decades ago [2]. The gas sensing material is generally deposited onto planar interdigitated electrodes (IDEs), or onto some functionally equivalent structure, like coils or meshes. The back side of the IDE is provided, in most cases, with a continuous electrode for heating, to allow setting the working temperature of the sensor for the gas to be detected.

There are many research groups worldwide working intensely in the field of gas sensing materials [2-11].

It is a challenge to find alternative gas sensing materials to the SnO₂, which has an operating temperature for methane as high as 600 °C [9-11]. The high value of the operating temperature is a serious drawback in the case of remote in-field applications, where energy supplies for heating the sensor might not be easily available. As such, room temperature (RT) sensing materials are highly desired.

The nanocomposite thin films have been prepared starting from SnSe₂ powder and Zn(II)-5,10,15,20-tetrakis-(4-aminophenyl)-porphyrin powder (ZnTAPP). These powders have been dissolved in amine- and methyl-containing solvents, ultrasonically mixed and dropcasted onto a sensor support with interdigital metal electrodes.

The methane sensing tests have been carried out at room temperature, using synthetic dry air, which alternatively contained or not, a flux of 1000 ppm of methane, for a period of 10 minutes for each alternation. The variation of the electrical resistance was of linear drop in the presence of methane and a corresponding linear increase, when only the synthetic dry air was present.

These nanocomposite thin films proved to be sensitive to methane in concentrations as low as 1000 ppm at room temperature of about 25 °C, without the need for heating the sensing element.

Acknowledgments

This research has been supported by the Romanian Ministry of Research and Innovation through the Core Program PN18-110101 and the project PN-III-2017-0172/15PCCDI, as well as by the Romanian Academy, through Research Programme Nr. 3/2018 at the Institute of Chemistry Timisoara.

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Light- and voltage-controlled capacity of Cu-As₂S₃-Cu thin film cells for photodetection

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Amorphous As₂S₃ thin films have remarkable electrical and optical properties, which make them solid candidates for a variety of applications and keep up the scientific interest to study them, even after decades from the first results [1]. This composition has its optical absorption edge around the 500 nm, and shows a high optical transmission in the infrared (IR) and mid-infrared (MIR) spectral domain [2]. From structural point of view, a-As₂S₃ has a quite sparse network, able to accommodate certain ionic or atomic species [2].

The behaviour of thin a-As₂S₃ film deposited onto a thin Ag layer, and illuminated with 532 nm wavelength green laser light has already been studied. These bi-layer structures change their structural and optical properties upon illumination with this monochromatic green light [2, 3]. In many technological solutions there is a tendency to substitute silver with copper.